

## YOZGAT BOZOK UNIVERSITY FACULTY OF ARTS AND SCIENCES CHEMISTRY DEPARTMENT COURSE PLAN

Course Code	Course Title	Semes ter	Course Type (C/E)	T+A+L (Time/Week)	Credi t	ECTS	Course Language		
KİM734	Organic Reaction Mechanisms	Spring	E	2+0+0	2	5	English		
	'	COURSE	INFORMAT	ION					
Course Catalog Description (Content)		Introduction to Organic Reaction Mechanisms, Inductive and Mesomeric Effect, Formal Charge and Oxidation Numbers, Introduction to Organic Reaction Mechanisms, Reaction Kinetics and Energy Diagrams, Basic Mechanisms, Nucleophilic Substitution Reactions (SN1, SN2, SNi), Effect of Adjacent Groups on Nucleophilic Substitution Reactions, Nucleophilic Substitution Reactions in Modern Synthesis, Elimination Reactions (E1, E2, E1kB), Elimination Reactions in Modern Synthesis, Addition Reactions, Conversion Reactions: Arrangement of Carbocations (Carbon central atom, Nitrogen, central atom, Oxygen central atom)-1, Conversion Reactions: Arrangements of Carbocations (Carbon central atom, Nitrogen central atom, Oxygen central atom)-2, Carbanion Regulations, Carbene Regulations, Azature Regulations Nitre Regulations, Radical Reactions							
The Aim of the Course		To emphasize the simple explanation and intelligibility of organic reaction mechanisms, To design original syntheses on reactions whose mechanisms have been learned, To offer solutions to the problems that may occur in synthesis reactions, to produce mechanisms for these reactions when an unknown reaction is encountered.							
Course L	evel	Degree							
Course L	anguage	English							
Teaching	method	(X) Formal ( ) Online ( ) Mixed/Hybrid							
Teaching Staff of the Course		Prof. Dr. Mustafa SAÇMACI Prof. Dr. Ş.Hakan ÜNGÖREN Prof. Dr. İrfan KOCA Dr. Öğr. Üyesi İbrahim Evren KIBRIZ							
Prerequis Course	site Course(s) of the	- -	Joi Ibramin L	VIOIT (ID) (IZ					
Learning Outcomes from the Course		Students will learn basic organic concepts and theories and will be able to explain organic mechanisms using this knowledge.							
		Students will learn the chemistry of the intermediate products formed during the formation of the organic product and will be able to analyze the product formation mechanisms.							
		Students will be able to bring an approach to explain the mechanism of a new organic reaction.							
		Students will understand advanced mechanisms and gain the ability to propose new mechanisms.							
		Students will gain the ability and skill to plan and research their own synthesis studies and to use the learned information							
	COURSE CONTENT								
Week T	heory		Pra	actice/Laboratory	1				



1	Introduction to Organic Reaction Mechanisms, Inductive and Mesomeric Effect, Formal Charge and Oxidation Numbers	
2	Introduction to Organic Reaction Mechanisms, Reaction Kinetics and Energy Diagrams	
3	Basic Mechanisms, Nucleophilic Substitution Reactions (SN1, SN2, SNi), Effect of Adjacent Groups on Nucleophilic Substitution Reactions	
4	Nucleophilic Substitution Reactions in Modern Synthesis	
5	Elimination Reactions (E1, E2, E1kB)	
6	Elimination Reactions in Modern Synthesis	
7	Conversion Reactions: Arrangement of Carbocations (Carbon central atom, Nitrogen central atom, Oxygen central atom)-1	
8	Midterm Exam	
9	Conversion Reactions: Arrangement of Carbocations (Carbon central atom, Nitrogen central atom, Oxygen central atom)-2	
10	Carbanion Regulations	
11	Carbene Arrangements	
12	Azuretic Arrangements	
13	Nitren Regulations	
14	Radical Reactions	
15	Final E	xam

## **Course Learning Resources**

Organic chemistry Jonathan Clayden, Nick Greeves, Stuart G Warren.
Organic Chemistry 7th Edition, T. W. Graham Solomons (Author), Craig B. Fryhle 1999.
Organic Reaction Mechanisms. Anac, O., Talinli N. (2008).
Reaction Mechanisms, Metin Balcı; Turkish Academy of Sciences Textbooks

ASSESSMENT CRITERIA				
Work Activities During the Semester	Number	Contribution		
Homework	1	%30		
Practice				
Forum/ Discussion Application				
Short Exam (Quiz)	2	%35		
Ratio Of Semester Studies To Semester Success (%)		%40		
Ratio of Final to Success (%)	1	%60		
Total		%100		

COURSE WORKLOAD TABLE				
Activity	Total Weeks	Duration (Weekly Hours)	Total Workload	
Midterm Examination	1	2	2	
Final Examination	1	2	2	
Attending Lectures	14	3	42	
Field Work	1	2	2	



Self St	tudy	14	3			42		
Individual Study for Mid term 7			4	4		28		
Individual Study for Final Examination			9	9		18		
Total V	Workload							
Total V	Workload / 25 (s)					136/25		
ECTS	Credits of the Course				1	36/25≌	 ≧5	
Note: T	he workload of the course will be o	letermined by the instructor on	a per-course ba	sis.				
	PROGRAM LE	EARNING OUTPUTS CONT	RIBUTION LE	VELS				
No	Program Learning Outputs		1	2	3	4	5	
1	Gains extensive knowledge about the basic chemical properties of matter and uses this knowledge in daily life, industrial scale, and practical chemistry and shares them with the society.						X	
2	Performs experiments, collects data, interprets, evaluates results, defines problems parallel to current technological developments, produces solutions against problems encountered in the laboratory.							
3	Calculates and processes chemical information and data.				Х			
4	Applies her/his knowledge and understanding of chemistry to the solution of unconventional qualitative and quantitative problems.					X		
5	Defines and comprehends chemical concepts and theories in Inorganic Chemistry, Organic Chemistry, Physical Chemistry, Analytical Chemistry, Biochemistry.						Х	
6	Can conduct research in the light of scientific data on any subject in the field of chemistry.					Х		
7	Writes, presents, discusses scientific material, and presents it orally to a knowledgeable audience.				Х			
8	makes environmental analyzes	Brings a chemical approach to the solution of environmental problems, makes environmental analyzes and reports.						
9	Knows a foreign language at a level to read and understand the basic terms and processes of the chemist profession.				Х			
10	Can use computer software and information and communication technologies at the level required by the field.			Х				
11	Adapts and transfers the knowledge gained in the field to secondary education.		-		Х			
12	Apart from the field of chemistry branches of science that she fee	els close to.		X				
13	awareness of taking responsibil	Carries out a study independently, makes group work and gains the awareness of taking responsibility.					Х	
14	They can develop a positive atticonstantly renew their profession	nal knowledge and skills.			X			
15	Have sufficient awareness of the justice, quality culture and prote protection, occupational health a	ction of cultural values, environ		X				

