

YOZGAT BOZOK UNIVERSITY FACULTY OF ARTS AND SCIENCES CHEMISTRY DEPARTMENT COURSE PLAN

2006									
Course Code	e Course Title	Semes ter	Course Type (C/E)	T+A+L (Time/Week)	Credi t	ECT S	Course Language		
KİM729	P Polymer Kinetics	1-2	E	2+0+0		4	Turkish		
		COURSE		TION	1	1			
Course Catalog Description (Content)		Reaction rate, Rate law, Reaction rate constant, Reaction mechanism, Molecularity, Half-life, Determination of reaction degrees, Pseudo degrees, Effect of temperature on reaction rate, Activation Energy, Sequential reactions, Theoretical determination of reaction rate, Evaluation of kinetic data, Catalysis and adsorption, Preliminary concepts, synthesis of polymers, additional concepts, polymers and small molecule substances, Cascading polymerization kinetics, Radical addition polymerization kinetics, Ionic polymerization kinetics, Copolymerization kinetics, Conductive polymer kinetics, Inputs used in the production of polymers are taught through theoretical lectures, examples and applications							
The Aim of the Course		To provide basic information about the kinetic analysis methods of polymers.							
Course	Level	Bachelor de	gree						
Course	Language	Turkish							
Teaching method		(X) Formal () Online () Mixed/Hybrid							
Teaching Staff of the Course		Prof. Dr. Ramazan COŞKUN, Prof. Dr. Ali DELIBAŞ, Asst. Prof. Dr. Hatice ARI							
Prerequisite Course(s) of the		-							
Learnir	ng Outcomes from the	1- Calculates reaction rate, molecularity, half-life and activation energy.							
Course		 2- Can determine the reaction degrees. 3- Calculate the kinetics of cascading, anionic, cationic, radical, etc. polymerizations. 4- Calculates copolymerization and conductive polymer kinetics. 5- Classifies the inputs used in the production of conductive polymers. 							
Week	Theory		P	Practice/Laboratory					
1	Basic Kinetic Concepts								
2	Determination of reaction degrees, Pseudo		Pseudo						
	degrees								
3	Effect of temperature on reaction rate, Activation Energy		ctivation						
4	Stepped Polymerization								
5	Polyesterization Kinetics								
6	Dependence of Polymerization Degree on Polymerization Amplitude		ee on						
7	Radical Addition Polymerization								

Radical Addition Polymerization Kinetics

Anionic Polymerization Kinetics

Cationic Polymerization Kinetics

Copolymerization Kinetics

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12	Conductive polymer kinetics		
13	Example Calculations		
14	An overview		
15	Final E	xam	

Course Learning Resources

1. Polimer Kimyası: Mehmet Saçak, Gazi Kitabevi, 2004, Ankara, Turkiye.

ASSESSMENT CRITERIA					
Work Activities During the Semester	Number	Contribution			
Homework	1	30			
Practice					
Forum/ Discussion Application					
Short Exam (Quiz)	2	70			
Ratio Of Semester Studies To Semester Success (%)		%40			
Ratio of Final to Success (%)	1	%60			
Total		%100			

COURSE WORKLOAD TABLE							
Activity	/	Total Weeks	Duration (Weekl Hours)	y Tot	Total Workload		
Theory		14	2		28		
Practic	e						
Forum/	Discussion Application						
Readin	g	14	2	28			
Interne	t Scanning, Library Study						
Individ	ual study	14	1		14		
Brainst	torming	14	1	14			
Presentation Preparation							
Presentation		011					
Final E	xam	1	1		1		
Prepara	ation for the Final Exam	3	5		15		
Diğer (Belirtiniz: Homework)							
Total Workload					100		
Total Workload / 25 (s)					100/25		
ECTS Credits of the Course					≌4		
Note: The workload of the course will be determined by the instructor on a per-course basis.							
PROGRAM LEARNING OUTPUTS CONTRIBUTION LEVELS						_	
NO	Program Learning Outputs		1 2	3	4	5	
1	1 Gains extensive knowledge about the basic chemical properties of matter and uses this knowledge in daily life, industrial scale, and practical chemistry and shares them with the society.				X		
2	2 Performs experiments, collects data, interprets, evaluates results, defines problems parallel to current technological developments, produces solutions against problems encountered in the laboratory				X		

4 Applies her/his knowledge and understanding of chemistry to the solution of unconventional qualitative and quantitative problems.

Calculates and processes chemical information and data.

3

Х

Х

5	Defines and comprehends chemical concepts and theories in		Y	
5	Dennes and comprehends chemical concepts and theories in		^	
	Inorganic Chemistry, Organic Chemistry, Physical Chemistry,			
	Analytical Chemistry, Biochemistry.			
6	Can conduct research in the light of scientific data on any subject in			Х
	the field of chemistry			
7	Writes presents discusses scientific material and presents it orally to			Y
1	whiles, presents, discusses scientific material, and presents it orally to			^
	a knowledgeable audience.			
8	Brings a chemical approach to the solution of environmental problems,		X	
	makes environmental analyzes and reports.			
9	Knows a foreign language at a level to read and understand the basic	X		
	terms and processes of the chemist profession			
10		X		
10	Can use computer software and information and communication	X		
	technologies at the level required by the field.			
11	Adapts and transfers the knowledge gained in the field to secondary	Х		
	education			
12	Apart from the field of chemistry, che/be gaine knewledge in different		v	
12	Apart from the field of chemistry, she/he gains knowledge in different		^	
	branches of science that she feels close to.			
13	Carries out a study independently, makes group work and gains the		X	
	awareness of taking responsibility.			
14	They can develop a positive attitude towards lifelong learning and			Х
	constantly renew their professional knowledge and skills			
45	Lieve sufficient successories of the universality of easiel rights assist	v		
15	Have sufficient awareness of the universality of social rights, social	X		
	justice, quality culture and protection of cultural values, environmental			
	protection, occupational health and safety.			

